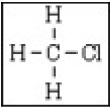
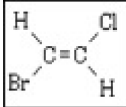
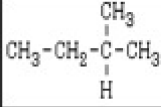
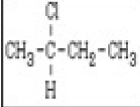
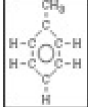


Nuclear/Organic Practice

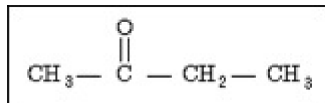
Multiple Choice

Identify the choice that best completes the statement or answers the question.

- When $^{214}_{84}\text{Po}$ decays, the emission consists consecutively of an alpha particle, then two beta particles, and finally another alpha particle. The resulting stable nucleus is
 - $^{206}_{83}\text{Bi}$
 - $^{210}_{83}\text{Bi}$
 - $^{206}_{82}\text{Pb}$
 - $^{208}_{82}\text{Pb}$
 - $^{210}_{81}\text{Tl}$
- The radioactive decay of $^{14}_6\text{C}$ to $^{14}_7\text{N}$ occurs by the process of
 - beta particle emission
 - alpha particle emission
 - positron emission
 - electron capture
 - neutron capture
- $^{251}_{98}\text{Cf} \rightarrow 2\text{n} + ^{131}_{54}\text{Xe} + \underline{\hspace{1cm}}$
 What is the missing product in the nuclear reaction represented above?
 - $^{114}_{42}\text{Mo}$
 - $^{118}_{44}\text{Ru}$
 - $^{120}_{42}\text{Mo}$
 - $^{120}_{44}\text{Ru}$
 - $^{122}_{46}\text{Pd}$
- The specific rate constant k for radioactive element X is 0.023 min^{-1} . What weight of X was originally present in a sample if 40. grams is left after 60. minutes?
 10. grams
 20. grams
 80. grams
 120. grams
 160. grams
- Correct statements about alpha particles include which of the following?
 - They have a mass number of 4 and a charge of +2.
 - They are more penetrating than beta particles.
 - They are helium nuclei.
 - I only
 - III only
 - I and II
 - I and III
 - II and III
- The nuclide $^{249}_{96}\text{Cm}$ is radioactive and decays by the loss of one beta particle. The product nuclide is
 - $^{245}_{94}\text{Pu}$
 - $^{245}_{95}\text{Am}$
 - $^{248}_{96}\text{Cm}$
 - $^{250}_{96}\text{Cm}$
 - $^{249}_{97}\text{Cm}$
- If 87.5 percent of a sample of pure ^{131}I decays in 24 days, what is the half-life of ^{131}I ?
 - 6 days
 - 8 days
 - 12 days
 - 14 days
 - 21 days
- Which of the following compounds exhibits optical isomerism?
 - 
 - 
 - 
 - 
 - 

- Which of the following pairs of compounds are isomers?
 - $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_3$ and $\text{CH}_3\text{-}\underset{\text{CH}_3}{\text{CH}}\text{-CH}_3$
 - $\text{CH}_3\text{-}\underset{\text{CH}_3}{\text{CH}}\text{-CH}_3$ and $\text{CH}_3\text{-}\underset{\text{CH}_3}{\text{CH}}\text{=CH}_2$
 - $\text{CH}_3\text{-O-CH}_3$ and $\text{CH}_3\text{-}\overset{\text{O}}{\parallel}\text{C-CH}_3$
 - $\text{CH}_3\text{-OH}$ and $\text{CH}_3\text{-CH}_2\text{-OH}$
 - CH_4 and $\text{CH}_2\text{=CH}_2$

10.



The organic compound represented above is an example of

- a. an organic acid c. an ether e. a ketone
b. an alcohol d. an aldehyde
11. Types of hybridization exhibited by the C atoms in propene, CH_3CHCH_2 , include which of the following?
- I. sp
II. sp^2
III. sp^3
- a. I only b. III only c. I and II only d. II and III only e. I, II, and III

**Nuclear/Organic Practice
Answer Section**

MULTIPLE CHOICE

1. ANS: C
2. ANS: A
3. ANS: B
4. ANS: E
5. ANS: D
6. ANS: E
7. ANS: B
8. ANS: D
9. ANS: A
10. ANS: E
11. ANS: D